

## Carbon-13 Nuclear Magnetic Resonance of 1,3,2-Dioxaphosphorinan-2-ones

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**Summary** The  ${}^3J({}^{13}\text{C}\dots{}^{31}\text{P})$  coupling constants in some 1,3,2-dioxaphosphorinan-2-ones have been shown to depend upon the POCC dihedral angle and the orientation of the P...O bond.

DETERMINATION of the  ${}^{13}\text{C}$ - ${}^{31}\text{P}$  coupling constants of organophosphorus compounds is useful for structural assignments.<sup>1</sup> However, the dependence of  ${}^{13}\text{C}$ - ${}^{31}\text{P}$  coupling

constants on the stereochemistry has not been previously reported.

The proton decoupled  ${}^{13}\text{C}$  n.m.r. spectra of a number of 1,3,2-dioxaphosphorinan-2-ones are given in Tables 1 and 2.

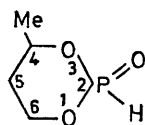
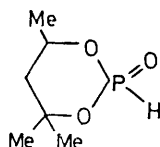
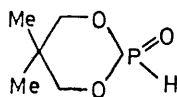
It has been shown by n.m.r. and magnetic dipole data<sup>2</sup> that in (**1a**) the P=O and Me groups are *cis* and equatorial and that the molecule is in the chair form, and in (**1b**) the P=O and Me groups are *trans* and the P=O group is axial.

TABLE 1

Compounds	$^{13}\text{C}$ chemical shifts (relative to $\text{Me}_4\text{Si}$ ), $\pm 0.2$ p.p.m.					
	C-4	C-5	C-6	4-Me <sup>eq</sup>	5-Me <sup>ax</sup>	6-Me <sup>eq</sup>
(1a)	74.4	33.7	65.9	22.9		
(1b)	74.8	32.7	64.6	22.4		
(2a)	71.8	45.1	83.3	32.3		27.0
(2b)	73.1	44.1	82.7	23.3		31.8
(3)	76.6	32.5	76.6		22.2	20.5

TABLE 2

Compounds	$^{13}\text{C}$ - $^{31}\text{P}$ coupling constants, $\pm 0.2$ Hz							
	$^2J_{\text{C-4}}$	$^3J_{\text{C-5}}$	$^2J_{\text{C-6}}$	$^3J_{\text{4-Me}^{\text{eq}}}$	$^4J_{\text{5-Me}^{\text{ax}}}$	$^4J_{\text{5-Me}^{\text{eq}}}$	$^3J_{\text{6-Me}^{\text{ax}}}$	$^3J_{\text{6-Me}^{\text{eq}}}$
(1a)	5.9	6.1	5.3	8.9				
(1b)	7.2	9.9	6.8	4.8				
(2a)	5.3	5.9	7.1	9.1			1.8	5.9
(2b)	6.4	11.0	7.1	6.0			<0.6	4.5
(3)	5.6	6.3	5.6		<0.3	1.0		

(1a) Me and P=O *cis*(1b) Me and P=O *trans*(2a) 4-Me and P=O *cis*(2b) 4-Me and P=O *trans*

(3)

The 5-methyl signals in the spectrum of compound (3) were assigned by selective  $^{13}\text{C}$ - $\{^1\text{H}\}$  decoupling.<sup>4</sup>

Tables 1 and 2 show the difference between the  $^{13}\text{C}$  n.m.r. data of isomers (1a) and (1b) and (2a) and (2b), especially the  $^{13}\text{C}$ - $^{31}\text{P}$  vicinal coupling constants, where the effect of the spatial arrangement of the P=O bond can be observed. Similar differences have recently been observed for *cis*- and *trans*-2-methyl-5-*t*-butyl-1,3,2-dioxaphosphorinans.<sup>5</sup> The  $^{13}\text{CH}_3(6)$ - $^{31}\text{P}$  constants demonstrate the dependence of the  $^{13}\text{C}$ - $^{31}\text{P}$  vicinal constants on the POCC dihedral angle. The constants are much higher for the equatorial methyls (dihedral angle, *ca.* 180°) than they are for the axial methyl (dihedral angle *ca.* 60°). This resembles the dependence of the  $^{31}\text{P}$ -H vicinal constants upon the dihedral angle in 1,3,2-dioxaphosphorinans.<sup>6</sup>

For compound (3), it is noteworthy that the long-range coupling constant  $^4J(^{13}\text{C}\dots^{31}\text{P})$  is observed only for the carbon of the equatorial methyl. Thus, as with the H-H coupling constants, the  $^{13}\text{C}$ - $^{31}\text{P}$  long-range coupling constants are greater when the interacting nuclei are *W* arranged (*W* rule).

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The configurations of (2a) and (2b) have similarly been determined.

The signals in the  $^{13}\text{C}$  spectra were assigned on the basis of the general trends for  $^{13}\text{C}$  chemical shifts<sup>3</sup> and by off-resonance partial proton decoupling.

<sup>1</sup> G. A. Gray, *J. Amer. Chem. Soc.*, 1971, **93**, 2132 and references therein.

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<sup>3</sup> E. F. Mooney and P. H. Winson, *Ann. Rev. N.M.R. Spectroscopy*, 1969, **2**, 153.

<sup>4</sup> K. D. Bartle, R. S. Edmundson, and D. W. Jones, *Tetrahedron*, 1967, **23**, 1701.

<sup>5</sup> W. G. Bentrude, K. C. Yee, R. D. Bertrand, and D. M. Grant, *J. Amer. Chem. Soc.*, 1971, **93**, 797.

<sup>6</sup> D. W. White and J. G. Verkade, *J. Magn. Resonance*, 1970, **3**, 111.